# SYNTHESIS OF ( $\pm$ )-7- AND 8-HYDROXYDUNNIONE 

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#### Abstract

The structures of two new natural products 7-and 8-hydroxydunnione have been confirmed by simple syntheses of their racemic forms.


Recently, two new naphthoquinones were isolated (1) among other quinones from Streptocarpus dunnii Mast. and have been identified mainly on spectral grounds as 7-and 8 -hydroxydunnione. The 7 -hydroxy compound has been obtained in the dextrorotary modification, whereas only the racemic form was observed in the second case. These substances are of particular interest because their biosynthesis has been shown to involve a unique mode of prenylation (2).

Expeditious syntheses of these natural products can be envisioned using an approach used in the instance of trypethelones (3). The required substrates, derivatives of 2,7- and 3,5-dihydroxynaphthoquinones are, in principle, readily accessible by application of simple regiospecific procedures devised earlier (4,5). Some 2,6-isomers were also prepared for purposes of identification and comparison. These compounds or their derivatives, when known, have previously been obtained only by multi-step processes involving the appropriate tetralone (6-8) or disubstituted naphthalene (9-11).

Cycloaddition of 1,3-bistrimethylsiloxy-1,3-butadiene (1) (12) to 2-chloro-5- or 6-methoxybenzoquinones ( $\mathbf{2}$ or 3 )(13-15) was followed by percolating a solution of the adduct through a column of silica gel and gave the corresponding naphthoquinones ( $4 \mathbf{4}$ and $\mathbf{5 a}$ ) in nearly quantitative yield (Scheme 1). The acetylated derivative, 7 -acetoxy-2methoxynaphthoquinone (4c), on ether exchange with silver (I) oxide and isoprenyl bromide in HMPA, followed by Claisen rearrangement in boiling EtOH and cyclization in cold concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}$, yielded 7-hydroxydunnione (7a), essentially in a oneflask procedure with an overall yield of $36 \%$ (along with a small amount of the rearranged $\beta$-isodunnione, $7 \mathbf{b}$ ).

5-Acetoxy-3-methoxynaphthoquinone ( $\mathbf{6 a}$ ) (16) was subjected to the same reaction conditions but afforded only $17 \%$ of 8 -hydroxydunnione (8a) (and $4 \%$ of the $\beta$-isocompound $\mathbf{8 b}$ ). However, when the corresponding diacetate ( $\mathbf{6 b}$ ) (9) was carried through the same steps, a $29 \%$ yield of the desired quinone was isolated (Scheme II).

Although the natural products appear to be unavailable, comparison of the exten-



Scheme II
sive data of the literature with those of the synthetic materials leaves no doubt as to their similarity.

## EXPERIMENTAL

All melting points were taken for samples in capillary tubes with a Thomas-Hoover Apparatus and are not corrected. The uv spectra were determined on a Hewlett-Packard 8450A spectrophotometer and the ir spectra on a Beckman Model IR-4250 instrument calibrated with a film of polystyrene. ${ }^{1} \mathrm{H}-\mathrm{nmr}$ spectra were recorded with a Varian XL-200 spectrometer using TMS as internal standard. Mass spectra were obtained with a Hewlett-Packard 5995A spectrometer. Merck silica gel $60 \mathrm{~F}_{254}$ for dry column chromatography was used throughout in a product to adsorbent ratio of 1:50-100. Elemental analyses were carried out by Galbraith Laboratories, Inc., Knoxville, Tenn. Exact masses were provided by the Laboratoire de spectrométrie de masse, Université de Sherbrooke, Sherbrooke, Québec.

Preparation of naphthoQuinones.-General method.-A solution of the benzoquinone ( 2.0 mmol ) and bistrimethylsiloxybutadiene (12) ( 2.5 mmol ) in dry $\mathrm{C}_{6} \mathrm{H}_{6}(10 \mathrm{ml})$ was heated to reflux for ca. 18 h , cooled, and poured onto a column of silica gel ( 60 g ). Slow elution with a mixture of $\mathrm{C}_{6} \mathrm{H}_{6}-\mathrm{ErOAc}$ (2:1) provided the corresponding naphthoquinone.

7-HYDROXY-2-METHOXYNAPHTHOQUINONE (4a).-A reaction carried out with 2-chloro-5methoxybenzoquinone (2) (13-15) gave $\mathbf{4 a}(99 \%)$, $\mathrm{mp} 246.0-246.5^{\circ}(\mathrm{MeOH})$; ir $v \max (\mathrm{KBr}) \mathrm{cm}^{-1} 3340$ $\mathrm{br}, 1670,1640,1613,1590$, and 1573; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 263$ (4.29), 289 (4.04), and 336 (3.34); ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-d_{6}\right) \delta 3.83\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{OCH}_{3}\right), 6.22(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.16(1 \mathrm{H}, \mathrm{dd}, J=8.6,2.5 \mathrm{~Hz}$, $6-\mathrm{H}), 7.30(1 \mathrm{H}, \mathrm{d}, J=2.5 \mathrm{~Hz}, 8-\mathrm{H})$, and $7.84(1 \mathrm{H}, \mathrm{d}, J=8.6 \mathrm{~Hz}, 5-\mathrm{H}) ; \mathrm{ms} m / z 204\left(\mathrm{M}^{+}\right)$. Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{8} \mathrm{O}_{4}$ : C, 64.71; H, 3.95. Found: C, 64.72; H, 4.10.

2,7-DIMETHOXYNAPhTHOQUINONE ( $\mathbf{4 b}$ ). -This derivative was prepared from $4 \mathbf{4}$ by the usual method ( $\mathrm{MeI}-\mathrm{Ag}_{2} \mathrm{O}-\mathrm{CHCl}_{3}$ ) and in nearly quantitative yield, $\mathrm{mp} 214.0-214.5^{\circ}\left(\mathrm{CHCl}_{3} / \mathrm{MeOH}\right)$ [lit. (6) $\operatorname{mp} 217-218^{\circ}$; ir $\nu \max (\mathrm{KBr}) \mathrm{cm}^{-1} 1680,1642,1609,1590$, and 1573 ; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon)$ $261(4.39), 288(4.12)$, and $333(3.47) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 3.90\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{OCH}_{3}\right), 3.95(3 \mathrm{H}, \mathrm{s}, 7-$ $\left.\mathrm{OCH}_{3}\right), 6.11(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.22(1 \mathrm{H}, \mathrm{dd}, J=8.6,2.5 \mathrm{~Hz}, 6-\mathrm{H}), 7.57(1 \mathrm{H}, \mathrm{d}, J=2.5 \mathrm{~Hz}, 8-\mathrm{H})$, and 8.03 $(1 \mathrm{H}, \mathrm{d}, J=8.6 \mathrm{~Hz}, 5-\mathrm{H})$.

7-ACETOXY-2-METHOXYNAPHTHOQUINONE (4c).-Acetylation of $\mathbf{4 a}\left(\mathrm{Ac}_{2} \mathrm{O}-\mathrm{H}_{2} \mathrm{SO}_{4}\right)$ afforded the acetate $\mathbf{4 c}(87 \%)$, $\mathrm{mp} 206.5-207.5^{\circ}\left(\mathrm{CHCl}_{3} /\right.$ petroleum ether, bp $\left.60-80^{\circ}\right)$; ir $\nu \max \left(\mathrm{KBr}^{2}\right) \mathrm{cm}^{-1} 1762$, $1679,1645,1607,1596$, and 1583 ; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 249(4.33), 272 \mathrm{sh}(4.13), 279$ (4.15),
and $328(3.49) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 2.35(3 \mathrm{H}, \mathrm{s}, 7-\mathrm{OAc}), 3.91\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{OCH}_{3}\right), 6.18(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.47$ ( $1 \mathrm{H}, \mathrm{dd}, J=8.5,2.4 \mathrm{~Hz}, 6-\mathrm{H}), 7.85(1 \mathrm{H}, \mathrm{d}, J=2.4 \mathrm{~Hz}, 8-\mathrm{H})$, and $8.13(1 \mathrm{H}, \mathrm{d}, J=8.5 \mathrm{~Hz}, 5-\mathrm{H})$; ms m$/ z$ $246\left(\mathrm{M}^{+}\right)$. Anal. calcd for $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{O}_{5}: \mathrm{C}, 63.42 ; \mathrm{H}, 4.09$. Found: C, 63.11; H, 4.30.

6-HYDROXY-2-METHOXYNAPHTHOQUINONE (5a).-Application of the general method to 2-chloro-6-methoxybenzoquinone (3) (13-15) gave naphthoquinone 5 a in nearly quantitative yield, mp $292-293^{\circ}(\mathrm{MeOH})$ [lit. (8) mp 285-295 ${ }^{\circ}$; ir $v \max (\mathrm{KBr}) \mathrm{cm}^{-1} 3330 \mathrm{br}, 1662,1648,1610,1584$, and 1570; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 268$ (4.21), 290 (4.19), and 332 (3.45); ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}-d_{6}\right) \delta 3.84$ $\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{OCH}_{3}\right), 6.26(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.12(1 \mathrm{H}, \mathrm{dd}, J=8.6,2.5 \mathrm{~Hz}, 7-\mathrm{H}), 7.27(1 \mathrm{H}, \mathrm{d}, J=2.5 \mathrm{~Hz}, 5-$ H ), and $7.89(1 \mathrm{H}, \mathrm{d}, J=8.6 \mathrm{~Hz}, 8-\mathrm{H})$.

2,6-Dimethoxynaphthoquinone (5b).- The diether $\mathbf{5 b}$ was obtained from $\mathbf{5 a}$ (MeI- $\mathrm{Ag}_{2} \mathrm{O}$ $\mathrm{CHCl}_{3}$ ) in nearly quantitative yield, $\mathrm{mp} 229-230^{\circ}\left(\mathrm{CHCl}_{3} / \mathrm{MeOH}\right.$ ) [lit. (6) mp 232-233${ }^{\circ}$; ir $v \max (\mathrm{KBr})$ $\mathrm{cm}^{-1} 1672,1650,1601,1592$, and 1572 ; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 265$ (4.27), 290 (4.23), and 328 (3.54); ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 3.90\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{OCH}_{3}\right), 3.95\left(3 \mathrm{H}, \mathrm{s}, 6-\mathrm{OCH}_{3}\right), 6.13(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.16(1 \mathrm{H}$, $\mathrm{dd}, J=8.6,2.5 \mathrm{~Hz}, 7-\mathrm{H}), 7.53(1 \mathrm{H}, \mathrm{d}, J=2.5 \mathrm{~Hz}, 5-\mathrm{H})$, and $8.08(1 \mathrm{H}, \mathrm{d}, J=8.6 \mathrm{~Hz}, 8-\mathrm{H})$.

6-ACETOXY-2-METHOXYNAPHTHOQUINONE (5c).-Acetylation of 5a( $\mathrm{Ac}_{2} \mathrm{O}-\mathrm{H}_{2} \mathrm{SO}_{4}$ ) gave a 95\% yield of the acetate $5 \mathrm{c}, \mathrm{mp} \mathrm{187.5-188.0}^{\circ}\left(\mathrm{CHCl}_{3} /\right.$ petroleum ether, bp $\left.60-80^{\circ}\right)$ [lit. (8) $\left.\mathrm{mp} 187^{\circ}\right]$; ir $\nu \max , ~$ $(\mathrm{KBr}) \mathrm{cm}^{-1} 1768,1687,1648,1637,1610,1598$, and 1578 ; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 251$ (4.27), $280(4.22)$, and $328(3.51) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 2.36(3 \mathrm{H}, \mathrm{s}, 6-\mathrm{OAc}), 3.91\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{OCH}_{3}\right), 6.19(1 \mathrm{H}, \mathrm{s}$, $3-\mathrm{H}), 7.44(1 \mathrm{H}, \mathrm{dd}, J=8.4,2.2 \mathrm{~Hz}, 7-\mathrm{H}), 7.80(1 \mathrm{H}, \mathrm{d}, J=2.2 \mathrm{~Hz}, 5-\mathrm{H})$, and $8.17(1 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}$, 8 -H).

Preparation of hydroxydunniones.-General method.-A mixture of the methoxynaphthoquinone ( 2.0 mmol ), freshly prepared $\mathrm{Ag}_{2} \mathrm{O}(3.6 \mathrm{~g} ; 16 \mathrm{mmol})$, and redistilled isoprenyl bromide ( $3.0 \mathrm{~g} ; 20 \mathrm{mmol}$ ) in dry HMPA ( 15 ml ) was stirred for 20 h , diluted with $\mathrm{Et}_{2} \mathrm{O}$ ( 300 ml ), filtered, washed several times with $\mathrm{H}_{2} \mathrm{O}$, dried, and evaporated ( 1.8 g of $\mathrm{Ag}_{2} \mathrm{O}$ and 1.5 g of isoprenyl bromide in the case of 2-acetoxynaphthoquinones). The residue was dissolved in absolute ErOH ( 50 ml ), refluxed for 20 h and again evaporated. Cold concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}(5 \mathrm{ml})$ was then added, the mixture was stirred for 10 $\min$ at $0^{\circ}$, poured into ice $\mathrm{H}_{2} \mathrm{O}(300 \mathrm{ml})$, and extracted with $\mathrm{Et}_{2} \mathrm{O}$.
( $\pm$ )-7-HYDROXYDUNNIONE (2,3,4,5-TETRAHYDRO-7-HYDROXY-2,3,3-TRIMETHYINAPHTHO 1,2 -b\}FURAN-4,5-DIONE) (7a).-Application of the foregoing method to quinone $4 \mathbf{4}$ and separation of the crude product by chromatography ( $\mathrm{C}_{6} \mathrm{H}_{6}$-EtOAc, 2:1) gave 7-hydroxydunnione ( 7 a ) ( 186 mg ; $36 \%$ ), mp $206^{\circ}\left(\mathrm{C}_{6} \mathrm{H}_{6}\right)$ [lit. (1) $\mathrm{mp}(+)$-isomer 217-219 ${ }^{\circ}$ ]; ir $v \max (\mathrm{KBr}) \mathrm{cm}^{-1} 3180 \mathrm{br}, 1700,1640,1630,1600$, 1550, 1500, 1414, 1303, 1219, 1158, 1090, 1060, 1029, and 822; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 270$ (4.47), $277(4.50), 303(3.79)$, and $496(3.31) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 1.27\left(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{CH}_{3}\right), 1.45(3 \mathrm{H}, \mathrm{s}, 3-$ $\left.\mathrm{CH}_{3}\right), 1.46\left(3 \mathrm{H}, \mathrm{d}, J=6.6 \mathrm{~Hz}, 2-\mathrm{CH}_{3}\right), 4.66(1 \mathrm{H}, \mathrm{q}, J=6.6 \mathrm{~Hz}, 2-\mathrm{H}), 7.13(1 \mathrm{H}, \mathrm{dd}, J=8.4,2.6 \mathrm{~Hz}, 8-$ $\mathrm{H}), 7.53(1 \mathrm{H}, \mathrm{brs}, 7-\mathrm{OH}), 7.54(1 \mathrm{H}, \mathrm{d}, J=8.4 \mathrm{~Hz}, 9-\mathrm{H})$, and $7.62(1 \mathrm{H}, \mathrm{d}, J=2.6 \mathrm{~Hz}, 6-\mathrm{H}) ; \mathrm{ms} m / \mathrm{z} 258$ $\left(\mathrm{M}^{+}\right)$. Anal. calcd for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{O}_{4}: \mathrm{C}, 69.76 ; \mathrm{H}, 5.46$. Found: $\mathrm{C}, 69.90 ; \mathrm{H}, 5.65$.

A second band consisted of ( $\pm$ )-7-hydroxy- $\beta$-isodunnione (2,3,4,5-tetrahydro-7-hydroxy-2,2,3trimethylnaphtho (1,2-b]furan-4,5-dione) ( 7 b ) ( 24 mg ; $5 \%$ ), mp 203-204 ${ }^{\circ}\left(\mathrm{Et}_{2} \mathrm{O} /\right.$ petroleum ether, bp $60-80^{\circ}$ ); ir $\nu \max (\mathrm{KBr}) \mathrm{cm}^{-1} 3135 \mathrm{br}, 1700,1642,1629,1610,1580,1566,1554,1505,1447,1303$, $1260,1118,1078$, and 801 ; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 269(4.38), 277(4.41), 298$ (3.80), and 496 (3.18); ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 1.27\left(3 \mathrm{H}, \mathrm{d}, J=7.0 \mathrm{~Hz}, 3-\mathrm{CH}_{3}\right), 1.49\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{CH}_{3}\right), 1.53\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{CH}_{3}\right)$, $3.19(1 \mathrm{H}, \mathrm{q}, J=7.0 \mathrm{~Hz}, 3-\mathrm{H}), 7.13(1 \mathrm{H}, \mathrm{dd}, J=8.3,2.5 \mathrm{~Hz}, 8-\mathrm{H}), 7.53(1 \mathrm{H}, \mathrm{d}, J=8.3 \mathrm{~Hz}, 9-\mathrm{H}), 7.62$ $(1 \mathrm{H}, \mathrm{d}, J=2.5 \mathrm{~Hz}, 6-\mathrm{H})$, and $8.15(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 7-\mathrm{OH})$. Hrms, calcd for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{O}_{4}: 258.0891$. Found: 258.0891 .
( $\pm$ )-8-HYDROXYDUNNIONE (2,3,4,5-TETRAHYDRO-6-HYDROXY-2,3,3-TRIMETHYLNAPHTHO[ 1,2 -b]FURAN-4,5-DIONE) (8a). -a) A similar reaction involving 5-acetoxy-3-methoxynaphthoquinone (6a) (16) afforded 8-hydroxydunnione (8a) ( $88 \mathrm{mg} ; 17 \%$ ) after chromatography ( $\mathrm{C}_{6} \mathrm{H}_{6} / \mathrm{ErOAc}, 5: 1$ ), mp 148.5-149.0 $0^{\circ}\left(\mathrm{MeOH}\right.$ ) [lit. (1) mp 151-152 ${ }^{\circ}$; ir $v \max (\mathrm{KBr}) \mathrm{cm}^{-1} 1645,1613,1588,1450,1401$, $1383,1310,1230,1155$, and 1035 ; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 238 \operatorname{sh}(4.19), 259$ (4.30), 291 (3.76), and $412(3.72) ;{ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{CDCl}_{3}\right) \delta 1.26\left(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{CH}_{3}\right), 1.44\left(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{CH}_{3}\right), 1.46(3 \mathrm{H}, \mathrm{d}, J=6.6 \mathrm{~Hz}, 2-$ $\left.\mathrm{CH}_{3}\right), 4.64(1 \mathrm{H}, \mathrm{q}, J=6.6 \mathrm{~Hz}, 2-\mathrm{H}), 7.10(1 \mathrm{H}, \mathrm{dd}, J=8.8,1.1 \mathrm{~Hz}, 7-\mathrm{H}), 7.19(1 \mathrm{H}, \mathrm{dd}, J=7.3,1.1$ $\mathrm{Hz}, 9-\mathrm{H}), 7.53(1 \mathrm{H}, \mathrm{dd}, J=8.8,7.3 \mathrm{~Hz}, 8-\mathrm{H})$, and $11.93(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{OH})$; ms $m / z 258\left(\mathrm{M}^{+}\right)$. Anal. calcd for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{O}_{4}$ : C, 69.76; H, 5.46. Found: C, 69.62; H, 5.71.

A slower-moving zone yielded ( $\pm$ )-8-hydroxy- $\beta$-isodunnione ( $2,3,4,5$-tetrahydro- 6 -hydroxy-2,2,3-trimethylnaphtho[1,2-b]furan-4,5-dione) (8b) ( $23 \mathrm{mg} ; \mathbf{4} \%$ ), mp $116-117^{\circ}$ (petroleum ether, bp $65-110^{\circ}$ ); ir $v \max (\mathrm{KBr}) \mathrm{cm}^{-1} 1638,1612,1586,1497,1449,1400,1367,1323,1225,1191$, and 1164; uv $\lambda \max (\mathrm{MeOH}) \mathrm{nm}(\log \epsilon) 239 \mathrm{sh}(4.26), 255(4.31), 290(3.86)$, and 414 (3.76); ${ }^{1} \mathrm{H} \mathrm{nmr}$ $\left(\mathrm{CDCl}_{3}\right) \delta 1.27\left(3 \mathrm{H}, \mathrm{d}, J=7.0 \mathrm{~Hz}, 3-\mathrm{CH}_{3}\right), 1.49\left(3 \mathrm{H}, \mathrm{s}, 2 \mathrm{CH}_{3}\right), 1.51\left(3 \mathrm{H}, \mathrm{s}, 2 \mathrm{CH}_{3}\right), 3.20(1 \mathrm{H}, \mathrm{q}$,
$J=7.0 \mathrm{~Hz}, 3-\mathrm{H}), 7.11(1 \mathrm{H}, \mathrm{dd}, J=8.4,1.1 \mathrm{~Hz}, 7-\mathrm{H}), 7.18(1 \mathrm{H}, \mathrm{dd}, J=7.3,1.1 \mathrm{~Hz}, 9-\mathrm{H}), 7.52(1 \mathrm{H}$, dd, $J=8.4,7.3 \mathrm{~Hz}, 8-\mathrm{H})$, and $11.95(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{OH})$. Hrms, calcd for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{O}_{4}: 258.0891$. Found: 258.0891 .
b) A reaction carried out with 3,5-diacetoxynaphthoquinone (6b) (9) gave a $29 \%$ yield of $\mathbf{8 a}$ ( 151 mg ) and $6 \%$ of $\mathbf{8 b}(32 \mathrm{mg})$.

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## LITERATURE CITED

1. K. Inoue, S. Ueda, H. Nayeshiro, and H. Inouye, Phytochemistry, 22, 737 (1983).
2. K. Inoue, S. Ueda, H. Nayeshiro, N. Moritome, and H. Inouye, Phytochemistry, 23, 313 (1984).
3. V. Guay and P. Brassard, J. Org. Chem., 49, 1853 (1984).
4. C. Brisson and P. Brassard, J. Org. Chem., 46, 1810 (1981).
5. L. Boisvert and P. Brassard, Tetrabedron Lett., 24, 2453 (1983).
6. T.R. Kasturi and T. Arunachalam, Can. J. Chem., 44, 1086 (1966).
7. A.C. Baillie and R.H. Thomson, J. Chem. Soc., (C), 2184 (1966).
8. G.R. Birchall and A.H. Rees, Can. J. Chem., 52, 610 (1974).
9. R.H. Thomson, J. Org. Chem., 13, 870 (1948).
10. L.F. Fieser and R.H. Brown, J. Am. Chem. Sor., 71, 3615 (1949).
11. Y.S. Tsizin and M.V. Rubtsov, J. Org. Chem. USSR, 4, 2141 (1968).
12. T. Ibuka, Y. Mori, and Y. Inubushi, Tetrabedron Lett., 3169 (1976).
13. L.C. Raiford and J.G. Lichry, J. Am. Chem. Sor., 52, 4576 (1930).
14. L. Asp and B. Lindberg, Acta Cbem. Scand., 4, 60 (1950).
15. I.S. Ioffe and A.F. Sukhina, Zh. Obshch. Kbim., 23, 295 (1953) [Cbem. Abstr., 48, 2640d (1954)].
16. J.C. Wilson, J.A. Benbow, J.H. Bowie, and R.H. Prager, J. Cbem. Soc. Perkin Trans. II, 498 (1978).

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